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On minimal representation of heterogeneous mass transfer for simulation and parameter estimation: Application to breakthrough curves exploitation

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Abstract

This paper deals with the problem of mass transfer representation within heterogeneous systems. For the purpose of their modelling from material balances, it is emphasised the existence of a well-suited change of state variables which leads to a representation containing a minimal number of parameters. The benefits of the proposed approach are illustrated by an example of parameter estimation performed from experimental breakthrough curves in chromatography. Experimental data concerning the adsorption of p-nitrophenol onto granular activated carbon are taken from [Chern, J., Chien, Y. (2002). Adsorption of nitrophenol onto activated carbon: isotherms and breakthrough curves. *Water Research 36*, 647–655]. This representation has several advantages: from a numerical point of view for simulation and also for parameter estimation since the number of parameter groups (macro-parameters) is minimal. In order to confirm the relevance of the change of coordinates that we propose, a structural identifiability study is also performed in some cases.

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1. Introduction

Parameter estimation is extensively used in chemical engineering to determine transport properties, kinetics as well as thermodynamic equilibrium constants (Jallut, Thomas, Touré, & Diard, 2001). The treatment of experimental data obtained from chromatographic columns is a typical example of such an approach. This technique is used to characterise liquid or gas phase adsorbents (Golshan-Shirazi & Guiochon, 1992; Gritti, Piatkowski, & Guiochon, 2002; Hufton & Danner, 1993a, 1993b; Jolimaitre, Tayakout-Fayolle, Jallut, & Ragil, 2001; Tondeur, Kabir, Luo, & Granger, 1996), soils (Chen & Wagenet, 1995) as well as polymers (Von Meien, Biscaia, & Nobrega, 1997). One can perform specific experiments to extract the values of some of the parameters. For example, the frontal technique allows to determine the equilibrium isotherm by performing a global mass balance of the adsorbate (Gritti & Guiochon, 2003). Axial dispersion coefficient as well as porosities can be estimated from inert tracer experiments (Hejtmanek & Schneider, 1993). The other approach consists in performing the estimation of all the parameters from only one experiment. Such an approach allows to minimize the sources of errors, to save time and to avoid numerous experiments (Altenhöner, Meurer, Strube, & Schmidt-Traub, 1997). However, as far as the number of parameters to estimate is greater, a structural analysis of the model has to be performed.

In this paper, we propose a way to derive models for chromatographic columns in order to avoid a possible overparametrization and we link this question to the structural identifiability of the parameters. This approach is applied to distributed parameter models. We restrict ourselves to the case of a single component experiment: an adsorbate is transported isothermally by a carrier and is supposed to be adsorbed in a solid phase. When it is introduced according to a step function from an equilibrium state, the response

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Nomenclature	
$a_{\rm p} = 3/R$ pellet volumetric surface area (m ² m ⁻³)	
A, B	parameters for the Redlich–Peterson equilibrium model $(mol m^{-3})^M$
C _f	adsorbate concentration in the mobile phase (mol m^{-3})
C^*	adsorbate concentration in a mobile phase at equilibrium with a solid (mol m ^{-3})
C_{f1}	adsorbate concentration in a moone phase at equinoritant which a solid (morning) adsorbate inlat concentration (mol m^{-3})
C_{in}	$\hat{f}(\cdot)$ vector fields
f(f, g(f), f(f)) vector fields $e^{eq}(f) = 0$, equilibrium relation (mol m ⁻³)	
g'() - h()	o equilibrium relation (morini)
n()	Ucesion metrix
п ;	index for the CSTP or the semple time
l	nidex for the CSTR of the sample time
$J(\theta)$	number of neremators to be estimated
к 1. 1./	number of parameters to be estimated aquivalent mass transfer coefficient in the LDE model $(m^2 c^{-1})$
$\kappa_{\rm p}, \kappa_{\rm p}$	equivalent mass transfer coefficient in the LDF model (in s)
	parameter for the Langmuir model $(m^3 mol^{-1})$
к <u>2</u> 1	parameter for the Langmun model (in more)
	column lengul (m)
N	number of CSTPs in the lumped parameters model
N	number of comple points
$\mathbf{D}_{\mathbf{p}}$	Poelot number
re	recter number adsorbate concentration within the adsorbant (mal m^{-3})
<i>q</i> <i>a</i>	initial adsorbate concentration within the adsorbent (mol m^{-3})
q_0^{*}	adsorbate concentration within the adsorbant in equilibrium with a fluid phase (mol m ^{-3})
q	volumetric flow rate through the column $(m^3 s^{-1})$
Q P	rellet mean radius (m)
л с	Student variable
S	sensitivity matrix
t S	time (s)
ı 11	input
<i>u</i>	interstitial fluid velocity (m s ^{-1})
r ro	state space vector and its initial value
л, л() V1	simulated output
y mod V-mass	simulated output
y meas	bed axial coordinate (m)
A	
Greek letters	
$\alpha, \beta, \gamma, \delta$ notations for the parameters in Eq. (11)	
ε_{f}	bed external porosity
θ, θ'	parameters vector
$\hat{ heta}$	estimation of θ
$\xi = z/L$	dimensionless axial coordinate
σ	variance
τ_1, τ_p, τ_r	$\tau'_{\rm p}, \tau''_{\rm p}, \tau_{\rm c}, \tau'_{\rm c}$ mass transfer time constants involved in the models (s ⁻¹)
-, r, þ, þ, -, c	

of the system is the breakthrough curve. The models that are used to extract parameters from experimental breakthrough curves are formulated in the state space from suitable material balances of the adsorbate, according to the general form (1):

$$\begin{cases} \dot{x} = \frac{\partial x}{\partial t} = f(x, u, \theta) \\ g^{eq}(x^*, x, \theta) = 0 \\ y(t) = h(x, u, \theta) \\ x(t = 0) = x_0(\theta) \end{cases}$$

(1)

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