Effect of anodes decoration with metal and metal oxides nanoparticles on pharmaceutically active compounds removal and power generation in microbial fuel cells

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1. Introduction

Pharmaceutically active compounds (PhACs), have been frequently detected in wastewater effluents, surface water and drinking water [1]. PhACs consist of pharmaceuticals, pesticides, and endocrine disrupting chemicals, etc [2]. PhACs may pose a significant threat to environmental ecosystem and human health, even at low concentrations [3]. For example, pharmaceuticals can damage endocrine systems and affect hormonal regulation in aquatic organisms and wildlife [4]. In addition, misuse and overuse of antibiotics may boost the growth of antibiotic-resistant bacteria and compromise the effectiveness of antimicrobial therapy [5]. Conventional biological treatment process generally cannot effectively remove PhACs from wastewater [6]. Therefore, it is of importance to find an alternative strategy to combat PhACs in wastewater and drinking water.

Bioelectrochemical systems (BES) are electrochemical hybrid systems which are regarded as a promising alternative for common biological wastewater treatment technology, as the energy recovered could offset partial energy consumption during the process [7]. Microbial fuel cell (MFC) is a typical BES which can remove organic pollutants via oxidization or reduction at anodes or cathodes. Recent studies have found that MFCs show the special potential to remove some PhACs possibly due to the specially enriched exoelectrogens on electrodes, etc. The problem how to establish an effective removal of atenolol, fluoxetine, norfluoxetine, trimethoprim and diphenhydramine were reported in the anode chamber of MFCs[9]. Many factors influence pollutants removal at anode electrodes, such as anode microbial composition and community, supporting materials and catalysts on electrodes, etc. The problem how to establish an effective anode electrode to enhance PhACs removal in MFCs has not been well studied up to date.

Bare carbon materials are generally used as the supporting material of anode electrodes in MFCs. Recent researchers have found that decoration of carbon material based anode with metal or metal oxides can...
change its electro-chemical behavior and influence microbial community enriched on anode [11–12]. For example, Sun et al. [13] found that anode decoration with gold increased power generation in MFCs because of the enhanced electrochemical performance of the anode and the accelerated growth of biofilm. Our previous study has found that anode decoration with biogenic nano-palladium (bio-Pd) increased power generation in MFCs due to the improved redox ability of electrode and the specially enriched exoelectrogens [11].

Palladium (Pd) is a well-known catalyst for reductive dehalogenation [14]. Pd can initiate dehalogenation for chlorinated organic compounds when charged with molecular hydrogen in the presence of hydrogen donors [15]. Therefore, Pd may also possess potential to catalyze the reduction of chlorinated PhACs. Besides Pd, manganese oxide (MnO₂) and magnetite (Fe₂O₄) could also be used as the catalyst to remove PhACs [14,16,17]. MnO₂, which has a reduction potential of 1.23 V, can directly oxidize some PhACs such as triclosan, chloroprene and codeine through adsorption and promoted electron transfer [18]. Fe₂O₄ can abiotically reduce PhACs due to the presence of Fe(II) species. Fe(III) species as end-products of reduction process can be further reduced to Fe(II) by electrons from anaerobic biodegradation process in MFCs [16]. Furthermore, the Fe(III)/Fe(II) couples can act as redox mediators in MFCs to enhance pollutants removal through an electron shuttling mechanism [11]. However, the problem whether these metal or metal oxides nanoparticles can be used for anode modification to enhance PhACs removal and power generation in MFCs has not been answered and deserves further study.

In this study, anode electrodes were modified with MnO₂, Fe₂O₄ and Pd nanoparticles and applied in MFCs aimed to enhance power generation and PhACs removal. Four PhACs, i.e. diclofenac (DCF), iohexol (IOX), ibuprofen (IBF) and carbamazepine (CBZ), were selected and studied here. Cyclic voltammograms (CVs) was conducted to explore the electro-catalytic properties of the metal or metal oxide modified electrodes. High-throughput sequencing technology was used to investigate the effects of anode decoration with MnO₂, Fe₂O₄ and Pd nanoparticles on the changes of the microbial community and structure of anode biofilms. This study may help to establish a novel strategy to increase recalcitrant PhACs removal and power generation in MFCs.

2. Materials and methods

2.1. Chemicals and materials

A non-wet proofed carbon cloth was used as the anode electrode (HeSen Co. Ltd, Shanghai, China). The carbon cloth was pretreated by acetone and high temperature prior to use. Na₂PdCl₄, FeCl₃·6H₂O and Mn(AC)₃ were purchased from Sinopharm Chemical Reagent Co. (Beijing, China). DCF, IOX, IBF and CBZ were purchased from Sigma-Aldrich (purity > 99%). HPLC grade acetonitrile and methanol were purchased from Mreda Technology Co. (Columbia, USA). All other reagents used in this study were of analytical grade if not noted otherwise.

2.2. Synthesis of MnO₂, Fe₂O₄ and Pd nanoparticles

Fe₂O₄ nanoparticles were synthesized according to the procedure previously described by Cao et al. [19]. FeCl₃·6H₂O and sodium dodecylbenzene sulfonate were mixed and dissolved in ethylene glycol to make Solution A. NaOH was dissolved in ethylene glycol to make Solution B. A homogeneous mixture of Solution A and Solution B was prepared through mixing under vigorous magnetic stirring and then it was transferred into a 100 mL Teflon-lined stainless-steel autoclave. The autoclave was tightly sealed and heated at 200 °C in an oven for 90 min. After cooling to room temperature, the precipitation was rinsed with deionized water and absolute alcohol repeatedly, and then dried at 60 °C for further use.

The MnO₂ nanoparticles were prepared according to a previous publication [20]. Briefly, a KMnO₄ aqueous solution was added to a Mn (Ac)₂ aqueous solution under vigorous magnetic stirring and further sonicated for 1 h for reaction. The final product of MnO₂ was rinsed with deionized water, dried at 60 °C, and then stored at 4 °C for later use.

Pd nanoparticles were prepared using Shewanella oneidensis MR-1 as the reducing bacteria, which can reduce Pd²⁺ to Pd⁰ on cell walls and in periplasmic spaces [21]. S. oneidensis MR-1 was cultured aerobically in Luria-Bertani (LB) medium at 28 °C and then collected through centrifugation at 3000g. Pd nanoparticles was produced by S. oneidensis MR-1 suspension in M9 medium, as described by Windt et al. [21]. The Pd associated biomass was burned at 500 °C for 1 h and Pd powder was obtained for later use.

2.3. Preparation of MnO₂, Fe₂O₄ and Pd modified electrodes

MnO₂, Fe₂O₄ and Pd nanoparticles were loaded to the surface of pre-treated carbon cloth (2.5 × 2.5 cm²) using 5% Nafion (DuPont Co., USA) as the binder. In brief, the prepared MnO₂ (Fe₂O₄ or Pd) nanoparticles were mixed with carbon black powder and added by Nafion and isopropanol to make a mixed paste. The paste was then spread to each side of the pretreated carbon cloth and dried at 121 °C. The amount of MnO₂, Fe₂O₄ and Pd onto the carbon cloth electrodes was 5 mg cm⁻². Only carbon black modified carbon cloth was used as the control anode in MFCs. An anode electrode made of non-conductive material-nonwoven cloth was also investigated in MFCs to mimic common biofilm carrier in the bioreactor. At least three repeatable samples were prepared for each type of above electrodes and used for the following experiments.

2.4. MFC setup and operation

Dual-chamber MFCs were used for the whole experiment, which were composed of an anode chamber and a cathode chamber separated by a proton exchange membrane (PEM) (Nafion 117, DuPont Co., USA). Each chamber had a working volume of 27 mL (3 cm × 3 cm × 3 cm, length × width × height). Five different types of anode electrodes were tested in the MFCs, i.e. MnO₂ nanoparticles modified electrode (MnO₂ electrode), Fe₂O₄ nanoparticles modified electrode (Fe₂O₄ electrode), Pd nanoparticles modified electrode (Pd electrode), the control electrode with only carbon black decoration (CB electrode), and the electrode made of non-conductive nonwoven cloth (NW electrode). The cathode electrode was carbon paper loaded with Pt catalyst (0.5 mg cm⁻²) on each side. Activated sludge collected from Yanjing wastewater treatment plant (Beijing, China) was inoculated to the anode chambers to enrich electricity generating bacteria on anode electrode. The anode chamber was filled with synthetic wastewater containing (per L, pH 7.0): CH₃COONa 1.64 g, NH₄Cl 0.31 g, KCl 0.13 g, Na₂HPO₄·12H₂O 10.32 g, NaH₂PO₄·2H₂O 3.32 g, a vitamin solution Na₂PQ₀·12H₂O 10.32 g, NaH₂PQ₀·12H₂O 3.32 g, a vitamin solution 5 mL and trace minerals mineral solution 12.5 mL [22]. The ionic strength of the anolyte and catholyte were 125 mM and 150 mM, respectively. When the MFCs with the different anode electrodes attained stable power generation, PhACs degradation in MFCs was investigated by adding DCF, IOX, IBF or CBZ (5 mg L⁻¹) to the above synthetic wastewater. The anode chambers were sparged with N₂ gas for 5 min to create anaerobic atmosphere before operation. The catholyte was 50 mM phosphate buffer solution (PBS, pH 7.4). All the MFCs ran at 28 ± 2 °C.

2.5. Measurements and calculations

The reactor voltage drop (V) across a 1000 Ω resistor connected between the cathode and the anode was measured by a data acquisition system (USB8253, RBH Co., China) and a personal computer. Current density (Iₒ) was calculated according to I = V/RA, where V is the voltage, R is the external resistor and A is the project surface area of anode.
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