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Optimization of pre-combustion capture for thermal power plants using Pinch Analysis

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ABSTRACT

Carbon dioxide emissions from the chimneys of thermal power plants create major environmental risks. Therefore, an important step toward reducing the emissions in these power plants can be the carbon dioxide pre-combustion capture process. In this paper, a 150 MW thermal steam cycle power plant fueled by bagasse was studied. The power plant has an efficiency of 32.74%, and emits 246.52 t/h carbon dioxide. First, the design and simulation of a suggested pre-combustion carbon dioxide capture process was outperformed. In this process, the amount of carbon dioxide separation and capture using mono ethanol amine (MEA) 30 wt% as solvent is 90%. In this condition, the mass flow of bagasse was increased about 60% to keep the plant efficiency constant. At the same time, the energy loss as a result of the addition of the carbon dioxide recovery unit was around 11%. The process was optimized through Pinch Analysis to reduce energy waste and fuel flow. Moreover, it was indicated that power plant efficiency could be increased around 8% by integrating the hot exhaust gases from the gasification unit with power plant boiler using a heat recovery steam generation (HRSG) unit. With this modification, bagasse consumption was decreased by 23%.

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1. Introduction

One of the most important issues facing researchers today, leading to significant global climate change is greenhouse gas emission to the atmosphere. Carbon dioxide (CO₂) is the most significant anthropogenic greenhouse gas, arising mainly out of power generation. Carbon dioxide capture and storage (CCS) is one of the measures required for the reduction of CO₂ emission. CO₂ capture processes can be divided into three general categories: (1) Post combustion capture, (2) oxy-fuel combustion, and (3) pre-combustion capture [1]. Many processes are available for separating carbon dioxide from gas mixtures based on physical absorption, chemical absorption, adsorption, membrane processes, etc. In the standard absorption process, flue gas in the absorber contacts with the lean solvent. The CO₂ is absorbed by the solvent, which is sent to the stripper and heated there to release the CO₂. Finally, the regenerated solvent is re-cycled to the absorber [2].

Several studies have indicated that high selectivity can be reached with absorption processes using chemical solvents, with

the result being the increased production of pure CO₂ [3]. Davidson [4] and Kothandaraman et al. [5] reported that recovery of carbon dioxide using chemical solvents was more practical. Comparing different solvents, they found that Mono Ethanol Amin is the best for carbon dioxide separation in terms of energy consumption and Techno-Economics. Addition of CCS processes to power plants increases the energy loss, reducing the produced net power by 10–15% [6]. In 2011, Kunze et al. [7] conducted a study using exergy analysis on an integrated gasification combined cycle (IGCC) power plant with a CCS unit, and found that much of the energy was wasted in the gasification unit. The energy loss caused by the addition of a CCS unit could be reduced by the employment of various methods. In the processes where solvents were used to absorb gas, the energy required for solvent recovery could be reduced by changing the type of solvent [8].

It is worth saying the total energy requirement in the reboiler can be reduced by improving the design of the solvent recovery unit. Besides, the use of the heat generated by CCS process in the steam cycle can contribute to the decreased energy loss resulting from CCS unit [9]. Romeo et al. [10] suggested that the optimal method is to extract saturated steam midway through the low-pressure section of the turbine. In this method, the lowest quality

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