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Original Research Paper

Analysis of multi-scale Ni particles generated by ultrasonic aided electrical discharge erosion in pure water

Yifan Liu, Kunlun Zhu, Xianglong Li*, Faming Lin, Yan Li

College of Manufacturing Science and Engineering, Sichuan University, Chengdu 610065, China

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ABSTRACT

Electrical discharge erosion is widely applied in the fabrication process of the metallic particles in liquids. The Ultrasonic aided electrical discharge erosion is based on the spark discharge in pure water. The synthesized colloids were classified in accordance with the nano size and micron size. The higher magnifications of morphology, chemical compositions, the crystal structure of the multi-scale particles were observed and analyzed by SEM, TEM, EDS, and XRD. It is verified that ultrasonic wave influenced the morphology of micro/nanoparticles and the roughness of inner and external surfaces of hollow micro-particles. Besides, based on results of EDS, XRD, and Quantitative phase analysis, it is confirmed that nickel oxide was detected only on the surface of microparticles but the nickel oxide was easily obtained when nanoparticles were formed. In addition, ultrasound wave affected the oxidation reaction in both scales but the reaction was remarkably enhanced on nanoparticles. The DLS and LPSA were used to measure the size distributions for the nano and micron scale, respectively. The D-Values of both conditions shown that the ultrasound has an enhanced effect on decreasing the size distribution in both scales.

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1. Introduction

In the past two decades, the requirement of micro/nano-scale metal particles has rapidly increased in many application areas such as semiconductors, solar cells, catalysts, 3-D printing (printing electronic devices and additive manufacturing) and powder metallurgy [1–4]. Generally, metal particles with the size of nano or micro size possess unique or improved physical, thermal, electrical and chemical properties compared with their corresponding bulk materials. Therefore, synthesis of multi-scale metal particles has drawn more attention in the past years [5,6]. Electrical discharge erosion is a physical method that has been developed to fabricate various metal particles in micro/nanoscale. Compared to the conventional chemical methods [7], electrical discharge erosion method is widely utilized due to its low-cost, eco-friendly, and time-saving properties. During the discharge erosion process, metallic particles are formed in different ambient cryogenic media like condensed gas (liquid nitrogen) [8], flowing gas [9] and liquid [10]. Generally, the synthesized metal particles have a wide particle size distribution ranging from a few tens of nanometers to several tens of micrometers. The multi-scale (nano- and micro- scale)

distribution of synthesized particles is highly influenced by the dielectric mediums, energy input and discharge time. Especially, the wide particle size distribution is easy to be observed when liquid serves as the working medium. Some researchers applied electrical discharge erosion method to produce metallic particles in different solutions. Dvornik [11] used spark erosion to fabricate nanostructured WC-Co particles in distilled water. M.R. Shabgard [12] synthesized nano-structured tungsten carbide (WC) powder using electro-discharge process with tungsten and graphite electrodes, which were submerged in two different dielectrics (kerosene and deionized water). R.K. Sahu et al. [13] employed micro-electrical discharge to prepare the copper micro/nanoparticles under in de-ionized water mixed the stabilizers like polyvinyl alcohol (PVA) and polyethylene glycol (PEG). Tseng et al. [14–16] fabricated the silver gold nanoparticles in different dielectric solutions using the spark erosion. Their devices for generation particles in liquids are based on the electrical discharge machining (EDM). In this process of EDM, low voltages (less than 200 V) and large currents (magnitude of tens Ampere) are applied to supply the energy and trigger the discharge process. M. Mardanian [17] used electrical discharge with high voltage (12 KV) to treat a mixture of copper, indium, and selenium powders to synthesize CIS nanoparticles in pure ethanol. In addition, ultrasonic technology was usually to assisting synthesis of particles. Chang [18] used arc

* Corresponding author.
E-mail address: lixianglong@scu.edu.cn (X. Li).

89 discharge assisted-ultrasonic vibration to obtain TiO₂ nanoparticles
 90 in deionized water. Ghomi [19] synthesized gold nanoparticles
 91 in deionized water and pure ethanol using ultrasonic-assisted
 92 spark discharge and revealed that the ultrasonic wave increased
 93 the shape uniformity of the nanoparticles and decreased their size.
 94 Liu et al. [20,21] also used the ultrasound-assisted spark discharge
 95 erosion to fabricate Ni micro-particles. We mainly focused on gener-
 96 ating Ni micro-particles in different liquids and investigated the
 97 influence of different electrical parameters and liquids on the par-
 98 ticle size distribution as well as purity of particles. Apart from
 99 these, the formation mechanism of different particle structures
 100 (mainly in the microscale) and rate of hollow structural Ni particles
 101 also were studied in these works. Although we found that the
 102 ultrasound provided a positive effect on narrowing the size distri-
 103 bution of micro-particles and improving the proportion of hollow
 104 particles, the drawback of these works [20,21] is that nanoparticles
 105 were neglected which suspended in the colloidal solution and were
 106 difficult to collect at the end of the process by using the magnet.

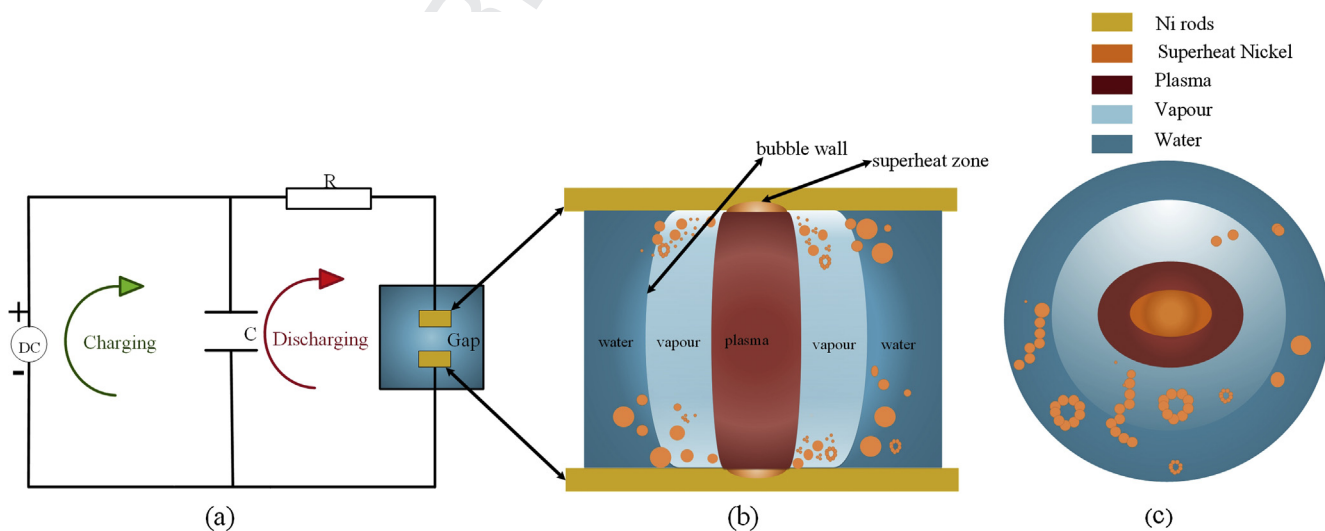
107 In this paper, we chose a new collection method which was
 108 described in Section 2. We pay more attention to the formation
 109 mechanism of different scale particles and explain the formation
 110 process through a high magnification observation like SEM (scan-
 111 ning electron microscopy) and TEM (transmission electron micro-
 112 scopy). The crystal structures of multi-scale particles were
 113 identified by XRD (X-ray diffraction) and SAED (selected area elec-
 114 tron diffraction). The chemical compositions were detected by EDS
 115 (energy dispersive spectroscopy). The size distributions of multi-
 116 scale particles were measured by Dynamic Light Scattering (DLS)
 117 for nanoscale and Laser Particle Size Analyzer (LPSA) for micron-
 118 scale.

119 **2. Experimental theory and methods**

120 The electric discharge generator is based on a controlled electri-
 121 cal discharge between two electrodes. The electrical circuit used on
 122 the electric discharge generator is equivalent to a simple RC circuit
 123 (Fig. 1a), where a capacitor placed in parallel to the gap between
 124 two sparking electrodes and charged by a constant current source
 125 with a low voltage (less than 200 V). When the electrode gap
 126 reaches a particular, very small size, the applied voltage on the
 127 capacitor exceeds the breakdown voltage of the dielectric solu-

128 tions. Then, the capacitor is discharged over the electrode gap
 129 and the dielectric molecule is ionized by the electric field, which
 130 results in an increase of the concentration of electrons and ions
 131 in the dielectric solutions between the spark gap. Subsequently,
 132 the conductive channel (plasma, coloring in red in Fig. 1b and c)
 133 is formed due to the high concentration of matters. The thermal
 134 energy enables local to superheat on the surface of electrodes (col-
 135 oring in orange in Fig. 1b and c), and the temperature rises locally
 136 to tens of thousands of degrees' Kelvin leading to heating, melting,
 137 boiling and evaporation of electrode material [22,23]. The dielec-
 138 tric liquid between the gap is simultaneously evaporated and
 139 molecules are dissociated into hydrogen and oxygen, resulting in
 140 a bubble (coloring in light blue in Fig. 1b and c) filled with gas
 141 and vaporized dielectric medium [24]. The molten drops and
 142 vaporized clusters are rapidly quenched and sintered to form par-
 143 ticles. When the plasma collapses at the end of the discharge, the
 144 superheated region boils violently and unstably, leading to ejecting
 145 metallic clusters and molten droplets (coloring in orange in Fig. 1b
 146 and c). The ejections pass straightly through the gas bubble and
 147 penetrate the bubble wall (sheath) into the dielectric liquid (col-
 148 oring in blue in Fig. 1b and c). The gap phenomena between the
 149 anode and the cathode regions can be simplified and described in
 150 Fig. 1(b) and (c).

151 It is noted that the inner pressure of the bubble is extremely
 152 high (280 Mpa) and the boundary between bubble and liquid
 153 expands with a velocity of several tens m/s [25,26]. The diam-
 154 eter of the bubble reaches several millimeters [23]. Obviously, there
 155 are opportunities to form different structures (spherical particles,
 156 non-spherical fractals, hollow particles, solid particles) as well as
 157 different scales due to the different phases of erosions and sur-
 158 rounding contents around the spark discharge zone. The different
 159 forming routes were described in our previous publication [21]
 160 through a hypothesis, i.e. the formation mechanisms were simply
 161 hypothesized by observing of visible morphology and structure
 162 of Ni particles. Actually, the difference in the morphology and scale
 163 of particles indicates the difference in the mechanisms of particles
 164 formation. Here, the different structures are interpreted on micro
 165 and nanoscale. Fig. 2 shows the schematic diagram of
 166 ultrasound-aided electric discharge erosion experimental process
 167 built to produce and analyze Ni multi-scale particles. The system
 168 consists of a power supply, a servo system, a pure water supply,
 169 ultrasound generator, a sample collector and a post-analyzing sys-



120 Fig. 1. Schematic representation of electric discharge erosion: (a) The spark as an RC circuit; (b) Schematic view of discharge gap (orange: Nickel; yellow: Nickel rods; dark
 121 red: plasma; light blue: vapor bubble; blue: dielectric solution); (c) Vertical view of discharge gap. (For interpretation of the references to colour in this figure legend, the
 122 reader is referred to the web version of this article.)

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